

Rh(III)-Catalyzed Cascade Annulation/C—H Activation of o-Ethynylanilines with Diazo Compounds: One-Pot Synthesis of Benzo[a]carbazoles via 1,4-Rhodium Migration

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Supporting Information

ABSTRACT: A Rh(III)-catalyzed cascade annulation/C-H activation of o-ethynylanilines with diazo compounds has been developed. This concise method allows for the rapid formation of a number of benzo[a] carbazoles in high yields, exhibiting good functional group tolerance and scalability. The key to the success of this approach involves one C-N bond and two C-C bond formation, and an aryl-to-aryl 1,4-rhodium migration.

B enzo[a] carbazoles are privileged motifs found in natural alkaloids and pharmaceutical molecules with anticancer and antiangiogenic activities, as well as in organic photovoltaic cells (OPVs), organic light-emitting diodes (OLEDs), and dye sensitized solar cells (DSSCs) due to their unique electronic and pronounced thermal stability properties (Figure 1).²

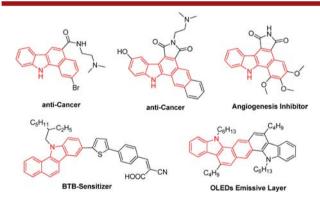


Figure 1. Alkaloids and photographic materials containing benzo [a]-carbazoles.

Generally, the benzo[a]carbazoles could be synthesized relying on the construction of ring A through photochemical cyclization, a palladium, indium, or copper-catalyzed cycloaromatization, intramolecular Friedel—Crafts arylation, Diels—Alder reaction, and sequential propargylation/cycloisomerization of preexisting indoles (Scheme 1a, left). On the other hand, the method to the synthesis of benzo[a]carbazoles also focused on the synthesis of ring B from amines or arylhydrazine hydrochlorides via iridium, palladium, or copper-catalyzed arylation/annulation, copper-promoted cycloaddition, and oxidative cyclo-dehydrogenation (Scheme 1a, right).

Scheme 1. Synthetic Approaches to Benzo[a]carbazoles

a) Previous work: approaches to benzo[a]carbazoles focus on the construction of single ring A or B

b) Previous work: cascade annulation of two alkynes to benzo[a]carbazoles

c) This work: cascade annulation/C-H activation of alkynes and diazo compounds to benzo[a]carbazoles

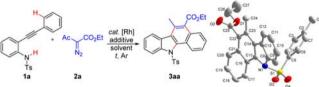
Compared to previous work that mostly focused on the construction of the single A ring or B ring to achieve benzo[a]carbazoles, the simultaneous synthesis of A and B rings has received increasing interest, because it not only brings greatly improved synthetic efficiency but also enhances aesthetic appeal for synthetic planning. Recently, Ohno¹¹ and Wu¹² disclosed gold-, palladium-catalyzed or iodine-mediated intramolecular cyclization of enediynes for the synthesis of benzo[a]carbazoles (Scheme 1b). Liang¹³ and Gong¹⁴ developed a palladium- and gold-catalyzed intermolecular tandem cyclization of two alkynes to synthesize benzo[a]-

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carbazoles (Scheme 1b). From the step economy as well as the functional group diversity point of view, the development of a new method to construct benzo[a]carbazoles utilizing easily prepared materials remains fascinating and challenging. Based on our continuous interest in the synthesis of nitrogen containing heterocycles, ¹⁵ herein, we will report a rhodium-catalyzed cascade annulation of o-ethynylanilines with diazo compounds to construct benzo[a]carbazoles. In this reaction, one C–N bond and two C–C bond formation was achieved with a one-pot procedure, and an aryl-to-aryl 1,4-rhodium migration was also involved (Scheme 1c).

Initially, we began our investigation with 4-methyl-*N*-(2-(phenylethynyl)phenyl)benzenesulfonamide **1a** and ethyl 2-diazo-3-oxobutanoate **2a** as model substrates (Table 1). A

Table 1. Optimization of Reaction Conditions^a



				OH OH	
entry	catalyst	additive	solvent	yield (%) ^b	
1	$[Cp*RhCl_2]_2$	_	DMF	59	
2	$[RhCl(COD)]_2$	_	DMF	nd^c	
3	$[Cp*RhCl_2]_2$	_	THF	<5	
4	$[Cp*RhCl_2]_2$	_	DMAC	70	
5	$[Cp*RhCl_2]_2$	$CuCl_2$	DMAC	33	
6	$[Cp*RhCl_2]_2$	CuO	DMAC	61	
7	$[Cp*RhCl_2]_2$	CuBr	DMAC	70	
8	$[Cp*RhCl_2]_2$	$Cu(OAc)_2$	DMAC	81	
9	$[Cp*RhCl_2]_2$	NaOAc	DMAC	66	
10	$[Cp*RhCl_2]_2$	KOAc	DMAC	45	
11 ^d	$[Cp*RhCl_2]_2$	$Cu(OAc)_2$	DMAC	91	
12 ^d	_	$Cu(OAc)_2$	DMAC	nd^c	
13 ^d	$[Cp*RhCl_2]_2$	_	DMAC	75	

"Reaction conditions: 1a (0.1 mmol), 2a (0.2 mmol), $[Cp*RhCl_2]_2$ (5 mol %), additives (30 mol %), solvent (1.0 mL), at room temperature under Ar atmosphere for 12 h, sealed tube. ^bIsolated yields. ^cnd = not determined. ^dThe reaction was conducted at 50 °C.

preliminary attempt with 5.0 mol % of [Cp*RhCl₂]₂ in 1.0 mL DMF at room temperature led to the benzo [a] carbazole 3aa in 59% yield, and the structure was unambiguously identified by single crystal X-ray analysis (entry 1). Replacing [Cp*RhCl₂]₂ with [RhCl(COD)]₂ resulted in no product (entry 2). A brief examination of solvents revealed that dipolar aprotic solvent DMAC was an ideal choice, which afforded 3aa in 70% yield (entry 4). The yield of 3aa increased to 81% in the presence of $Cu(OAc)_2$ when several copper salts were tested (entries 5–8). Considering the anion may play an important role in the reaction efficiency, several acetates were then tested and lower yields were obtained compared with $Cu(OAc)_2$ (entries 9–10). By increasing the temperature to 50 °C, the yield could be further increased to 91% (entry 11). Control experiments indicated that [Cp*RhCl₂]₂ was indispensable to this reaction (entries 12-13).

With the optimized conditions in hand, we then investigated the substrate scope of the *o*-ethynylanilines and diazo compounds to test the generality of this cascade cyclization, and the results are summarized in Scheme 2. The presence of methyl and methoxyl groups at C4- or C5-positions of the

Scheme 2. Substrate Scope a,b

"Reaction conditions: **1a** (0.1 mmol), **2a** (0.2 mmol), [Cp*RhCl₂]₂ (5 mol %), Cu(OAc)₂ (30 mol %), DMAC (1.0 mL), at 50 °C under Ar for 12 h, sealed tube. ^bIsolated yields. ^c3.0 mmol scale.

aniline ring afforded 3ba, 3da, and 3ea in 91%, 81%, and 71% yields. The halogen and electron-withdrawing groups delivered 3ca, 3fa-ia in 46-79% yields. The phenyl ring bearing electron-donating and -withdrawing groups at the meta- and para-positions performed smoothly and afforded 3ka and 3laqa in moderate to good yields. Notably, to increase product diversity, a heteroaryl-substituted substrate was also attempted. 4-Methyl-*N*-(2-(thiophen-2-ylethynyl)phenyl)-benzenesulfonamide 1r gave 3ra in 20% yield. When the ester group of 2a was altered from ethyl to methyl, tert-butyl, isopropyl, and benzyl groups, 3ab-ae were obtained in 82-92% yields. By changing the R³ group to *n*-propyl, cyclopropyl, and cyclohexyl groups, 3af, 3ag, and 3ah were prepared in 84%, 87%, and 50% yield, respectively. Ethyl 2-diazo-3-oxo-3-phenylpropanoate 2i offered 3ai in 70% yield. When 3-diazopentane-2,4-dione 2j was tested, the desired product 3aj was generated in 59% yield. Unfortunately, no corresponding product could be obtained when 2-diazo-1-morpholinobutane-1,3-dione 2k was used. To improve the practicability of this reaction, a gram-scale reaction was conducted and 3aa was isolated in 94% yield (1.29 g).

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With the aim to further evaluate the practicability of this reaction, late-stage modification of 3aa was conducted and is shown in Scheme 3. The sulfamide and ester group in 3aa

Scheme 3. Derivatization of Benzo[a]carbazole 3aa

could be hydrolyzed, and the corresponding products 4 and 5 were obtained in 96% and 86% yields. After the hydrogenation of 3aa, compound 6 with a hydroxyl group was obtained in 93% yield. Besides, 6 could react with DPPA to provide the benzo[a] carbazole 7 in 98% yield.

To gain some mechanistic insight into this transformation, control experiments and deuterated labeling experiments were carried out. 3aa could be obtained in 75% yield when $[Cp*RhCl_2]_2$ was used as the sole catalyst, while only single cyclic product 2-phenyl-1-tosyl-1H-indole 8 could be detected with $Cu(OAc)_2$ as the sole catalyst (Scheme 4a). Considering

Scheme 4. Mechanistic Experiments

that indole 8 was the possible intermediate for this reaction, we then examined the reaction between 8 and ethyl 2-diazo-3-oxobutanoate 2a under the standard conditions (Scheme 4b); no desired product 3aa was observed. These results suggested that indole 8 as an intermediate was not involved in this transformation. Moreover, the reaction was exclusively catalyzed by Rh(III) active species and Cu(OAc)₂ did not take part in the catalytic cycle, but just helped to generate

Rh(III) active species. In order to further illustrate the 1,4-rhodium migration process, a deuterated labeling experiment with $[D]_5$ -1a was conducted under the standard conditions (Scheme 4c). The observed incorporation of hydrogen at the C2 of indole 8 indicated the occurrence of 1,4-rhodium migration (see Supporting Information (SI) for details). Additionally, we examined the reaction of 1a in a 5:1 mixture of DMAC/D₂O solvent (Scheme 4d). The presence of D₂O provided deuterated $[D]_n$ -8 in 57% yield (see SI for details). This result suggested that the aryl-to-aryl 1,4-Rh(III) migration occurred in a concerted metalation—deprotonation/reprotonation sequence (II to IV in Scheme 5).

Scheme 5. Proposed Reaction Mechanism

Based on the above results and literature precedents, 16-20 a plausible reaction pathway of this Rh(III)-catalyzed cascade annulation/C-H activation of o-ethynylanilines with diazo compounds was proposed and is shown in Scheme 5. Coordination of an alkyne to active Rh(III) species affords intermediate I, 18 which could increase the electrophilicity of the triple bond. The nucleophilic attack of the nitrogen on the triple bond generates the benzoheterocyclic Rh(III) species II. 19 An acetate promoted, concerted metalation-deprotonation of II gives rhodacycle III, which could undergo acetolysis to give IV. 16 Subsequently, diazo compound 2a reacts with intermediate IV to generate Rh-carbene intermediate V with the release of N₂ and Rh-carbene migratory insertion produces intermediate VI. 20 Upon protonation, intermediate VII could be achieved along with the regeneration of the active rhodium(III) species. Finally, intramolecular dehydration condensation of VII yields the product 3aa.

In conclusion, we have developed a novel intermolecular cascade annulation/C—H activation of o-ethynylanilines with diazo compounds driven by the rhodium catalyst. The benzo[a] carbazoles were obtained in high yields, exhibiting good functional group tolerance and scalability. The reaction proceeds through a catalytic cycle involving intramolecular nucleophilic addition of o-ethynylanilines, aryl-to-aryl 1,4-

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rhodium migration. In this strategy, one C-N bond and two C-C bond formation was achieved with a one-pot procedure.

ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.6b02534.

¹H and ¹³C NMR spectra for all new compounds (PDF) X-ray crystallographic data for **3aa** (CIF)

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Notes

The authors declare no competing financial interest.

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REFERENCES

- (1) For selected examples, see: (a) Knölker, H. J.; Reddy, K. R. Chem. Rev. 2002, 102, 4303. (b) Peifer, C.; Stoiber, T.; Unger, E.; Totzke, F.; Schächtele, C.; Marmé, D.; Brenk, R.; Klebe, G.; Schollmeyer, D.; Dannhardt, G. J. Med. Chem. 2006, 49, 1271. (c) Routier, S.; Mérour, J.-Y.; Dias, N.; Lansiaux, A.; Bailly, C.; Lozach, O.; Meijer, L. J. Med. Chem. 2006, 49, 789. (d) Wang, Y.-Q.; Li, X.-H.; He, Q.; Chen, Y.; Xie, Y.-Y.; Ding, J.; Miao, Z.-H.; Yang, C.-H. Eur. J. Med. Chem. 2011, 46, 5878. (e) Schmidt, A. W.; Reddy, K. R.; Knölker, H.-J. Chem. Rev. 2012, 112, 3193.
- (2) For selected examples, see: (a) Qian, X.; Zhu, Y.-Z.; Chang, W.-Y.; Song, J.; Pan, B.; Lu, L.; Gao, H.-H.; Zheng, J.-Y. ACS Appl. Mater. Interfaces 2015, 7, 9015. (b) Paramasivam, M.; Chitumalla, R. K.; Singh, S. P.; Islam, A.; Han, L.; Rao, V. J.; Bhanuprakash, K. J. Phys. Chem. C 2015, 119, 17053. (c) Ivaniuk, K.; Cherpak, V.; Stakhira, P.; Hotra, Z.; Minaev, B.; Baryshnikov, G.; Stromylo, E.; Volyniuk, D.; Grazulevicius, J. V.; Lazauskas, A.; Tamulevicius, S.; Witulski, B.; Light, M. E.; Gawrys, P.; Whitby, R. J.; Wiosna-Salyga, G.; Luszczynska, B. J. Phys. Chem. C 2016, 120, 6206.
- (3) (a) Protti, S.; Palmieri, A.; Petrini, M.; Fagnoni, M.; Ballini, R.; Albini, A. Adv. Synth. Catal. 2013, 355, 643. (b) Li, Y.; Pang, Z.; Zhang, T.; Yang, J.; Yu, W. Tetrahedron 2015, 71, 3351.
- (4) (a) Tsuchimoto, T.; Matsubayashi, H.; Kaneko, M.; Nagase, Y.; Miyamura, T.; Shirakawa, E. J. Am. Chem. Soc. 2008, 130, 15823. (b) Shi, Z.; Ding, S.; Cui, Y.; Jiao, N. Angew. Chem., Int. Ed. 2009, 48, 7895. (c) Xie, R.; Ling, Y.; Fu, H. Chem. Commun. 2012, 48, 12210. (d) Bhunia, S. K.; Polley, A.; Natarajan, R.; Jana, R. Chem. Eur. J. 2015, 21, 16786.
- (5) Zhou, J.; Yang, W.; Wang, B.; Ren, H. Angew. Chem., Int. Ed. **2012**, 51, 12293.
- (6) (a) Tao, Y.; Zhang, F.; Tang, C.-Y.; Wu, X.-Y.; Sha, F. Asian J. Org. Chem. **2014**, 3, 1292. (b) Sha, F.; Tao, Y.; Tang, C.-Y.; Zhang, F.; Wu, X.-Y. J. Org. Chem. **2015**, 80, 8122.
- (7) Reddy, C. R.; Valleti, R. R.; Dilipkumar, U. Chem. Eur. J. 2016, 22, 2501.
- (8) (a) Cá, N. D.; Sassi, G.; Catellania, M. Adv. Synth. Catal. 2008, 350, 2179. (b) Zhu, D.; Liu, Q.; Luo, B.; Chen, M.; Pi, R.; Huang, P.;

Wen, S. Adv. Synth. Catal. 2013, 355, 2172. (c) Kale, A. P.; Kumar, G. S.; Mangadan, A. R. K.; Kapur, M. Org. Lett. 2015, 17, 1324. (d) Suzuki, C.; Hirano, K.; Satoh, T.; Miura, M. Org. Lett. 2015, 17, 1597.

- (9) Wang, J.; Li, J.; Zhu, Q. Org. Lett. 2015, 17, 5336.
- (10) (a) Xiao, F.; Liao, Y.; Wu, M.; Deng, G.-J. *Green Chem.* **2012**, 14, 3277. (b) Truax, N. J.; Mejia, F. B.; Kwansare, D. O.; Lafferty, M. M.; Kean, M. H.; Pelkey, E. T. *J. Org. Chem.* **2016**, 81, 6808.
- (11) (a) Hirano, K.; Inaba, Y.; Watanabe, T.; Oishi, S.; Fujii, N.; Ohno, H. Adv. Synth. Catal. 2010, 352, 368. (b) Hirano, K.; Inaba, Y.; Takahashi, N.; Shimano, M.; Oishi, S.; Fujii, N.; Ohno, H. J. Org. Chem. 2011, 76, 1212. (c) Hirano, K.; Inaba, Y.; Takasu, K.; Oishi, S.; Takemoto, Y.; Fujii, N.; Ohno, H. J. Org. Chem. 2011, 76, 9068.
- (12) (a) Chen, C.-C.; Chin, L.-Y.; Yang, S.-C.; Wu, M.-J. Org. Lett. **2010**, 12, 5652. (b) Chen, C.-C.; Yang, S.-C.; Wu, M.-J. J. Org. Chem. **2011**, 76, 10269.
- (13) Xia, X.-F.; Wang, N.; Zhang, L.-L.; Song, X.-R.; Liu, X.-Y.; Liang, Y.-M. J. Org. Chem. **2012**, 77, 9163.
- (14) Li, N.; Lian, X.-L.; Li, Y.-H.; Wang, T.-Y.; Han, Z.-Y.; Zhang, L.; Gong, L.-Z. Org. Lett. **2016**, 18, 4178.
- (15) (a) Zhao, L.; Li, Z.; Chang, L.; Xu, J.; Yao, H.; Wu, X. Org. Lett. 2012, 14, 2066. (b) Su, Y.; Zhou, H.; Chen, J.; Xu, J.; Wu, X.; Lin, A.; Yao, H. Org. Lett. 2014, 16, 4884. (c) Jiang, H.; Gao, S.; Xu, J.; Wu, X.; Lin, A.; Yao, H. Adv. Synth. Catal. 2016, 358, 188. (d) Sun, P.; Wu, Y.; Yang, T.; Wu, X.; Xu, J.; Lin, A.; Yao, H. Adv. Synth. Catal. 2015, 357, 2469. (e) Li, X.; Gai, K.; Yuan, Z.; Wu, J.; Lin, A.; Yao, H. Adv. Synth. Catal. 2015, 357, 3479. (f) Wu, Y.; Sun, P.; Zhang, K.; Yang, T.; Yao, H.; Lin, A. J. Org. Chem. 2016, 81, 2166. (g) Sun, P.; Wu, Y.; Huang, Y.; Wu, X.; Xu, J.; Yao, H.; Lin, A. Org. Chem. Front. 2016, 3, 91.
- (16) For selected examples of 1,4-rhodium migration reactions, see: (a) Ma, S.; Gu, Z. Angew. Chem., Int. Ed. 2005, 44, 7512. (b) Seiser, T.; Roth, O. A.; Cramer, N. Angew. Chem., Int. Ed. 2009, 48, 6320. (c) Souillart, L.; Cramer, N. Chem. Sci. 2014, 5, 837. (d) Burns, D. J.; Lam, H. W. Angew. Chem., Int. Ed. 2014, 53, 9931. (e) Burns, D. J.; Best, D.; Wieczysty, M. D.; Lam, H. W. Angew. Chem., Int. Ed. 2015, 54, 9958. (f) Korkis, S. E.; Burns, D. J.; Lam, H. W. J. Am. Chem. Soc. 2016, 138, 12252.
- (17) For selected examples, see: (a) Yamabe, H.; Mizuno, A.; Kusama, H.; Iwasawa, N. J. Am. Chem. Soc. 2005, 127, 3248. (b) Partridge, B. M.; González, J. S.; Lam, H. W. Angew. Chem., Int. Ed. 2014, 53, 6523. (c) Wang, M.; Zhang, X.; Zhuang, Y.-X.; Xu, Y.-H.; Loh, T.-P. J. Am. Chem. Soc. 2015, 137, 1341. (d) Hu, T.-J.; Zhang, G.; Chen, Y.-H.; Feng, C.-G.; Lin, G.-Q. J. Am. Chem. Soc. 2016, 138, 2897
- (18) For selected examples, see: (a) Seoane, A.; Casanova, N.; Quiñones, N.; Mascareñas, J. L.; Gulías, M. J. Am. Chem. Soc. 2014, 136, 834. (b) Zheng, J.; Zhang, Y.; Cui, S. Org. Lett. 2014, 16, 3560. (c) Park, J.; Mishra, N. K.; Sharma, S.; Han, S.; Shin, Y.; Jeong, T.; Oh, J. S.; Kwak, J. H.; Jung, Y. H.; Kim, I. S. J. Org. Chem. 2015, 80, 1818. (19) For selected examples, see: (a) Mizukami, A.; Ise, Y.; Kimachi,
- (19) For selected examples, see: (a) Mizukami, A.; Ise, Y.; Kimachi, T.; Inamoto, K. *Org. Lett.* **2016**, *18*, 748. (b) Hu, Z.; Tong, X.; Liu, G. *Org. Lett.* **2016**, *18*, 2058.
- (20) For selected examples of carbene migratory insertion, see: (a) Shi, Z.; Koester, D. C.; Boultadakis-Arapinis, M.; Glorius, F. *J. Am. Chem. Soc.* 2013, 135, 12204. (b) Xia, Y.; Liu, Z.; Liu, Z.; Ge, R.; Ye, F.; Hossain, M.; Zhang, Y.; Wang, J. *J. Am. Chem. Soc.* 2014, 136, 3013. (c) Yu, S.; Liu, S.; Lan, Y.; Wan, B.; Li, X. *J. Am. Chem. Soc.* 2015, 137, 1623. (d) Zhao, D.; Kim, J. H.; Stegemann, L.; Strassert, C. A.; Glorius, F. *Angew. Chem., Int. Ed.* 2015, 54, 4508. (e) Chen, X.; Hu, X.; Bai, S.; Deng, Y.; Jiang, H.; Zeng, W. *Org. Lett.* 2016, 18, 192.